

LIMITS OF SEPARATION PERFORMANCE IN SFC

SHIMADZU SFC USER MEETING

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- Supercritical Fluid Chromatography/SFC (incl. sub-critical conditions) employs a low viscosity mobile phase that contains mainly CO₂
 - Viscosity η between 0.1-0.25 cP for 10-50% MeOH at 30°C and 150 bar (vs. 0.8 cP for water at 30°C) (1 cP = 1mPa·s)
 - → Fast separations possible: high $D_{mol} \sim 1/\eta$
 - Optimal velocity u_{opt} ~ D_{mol}
 - Fast mass transfer: C-term ~ 1/D_{mol}

→ Low column pressure drop since $\Delta P \sim \eta$ (but also $\sim u_{opt}$)

• "Green" technique (low waste/solvent use), unique selectivity and orthogonality, chiral separations, advantages for scale up to prep...





Supercritical Fluid Chromatograph

SFC Basic Guide



Fig. 8 Linear Velocity vs. Column Back Pressure





Fig. 1 Phase Diagram for Carbon Dioxide

Table 1 Physical Properties of Supercritical Fluids

	Diffusion Coefficient (m ² /s)	Density (g/cm³)	Viscosity (g/cm·s)
Liquids	10-6	1	10-2
Supercritical Fluids	10-3	0.2-0.8	10-3
Gases	10-1	10-3	10-4



- Diffusion coefficients in SFC claimed to be very high (10⁻⁷-10⁻⁸ m²/s) relative to LC (10⁻⁹ m²/s)
- Range of D_{mol} based on small neutral molecules in pure CO_2 at low backpressure (< 100 bar)
- Addition of minor amounts of co-solvent strongly decrease D_{mol}





T. Januarius et al., J. Chromatogr. A (2022) 463485

- Density ρ of SFC mobile phases similar or higher than that of modifier







Figure 2.1 Density of pure CO_2 as a function of pressure at four different temperatures. From top to bottom: 40, 60, 80 and 100 °C.



T. Januarius et al., J. Chromatogr. A (2022) 463485

LIMITS OF SEPARATION POWER IN CHROMATOGRAPHY

- Time-efficiency trade off is given by the Knox-Saleem equation for a fully optimized system (flow rate, particle size and column length)
- "Money-equation":
 - higher separation efficiency N is valuable/expensive



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LIMITS OF SEPARATION POWER: LC VS. SFC

- SFC: same columns as in LC (h_{min} and ϕ)
- Viscosity is 3-8x lower in SFC than LC
- What about pressure drop?



Mode	P _{max} (bar)	F (ml/min)
LC	1300-1500	2-5
SFC	413-660	3-5

• SFC allows fast separations at low pressure drops, but has similar kinetic performance limits than UHPLC



LIMITS OF SEPARATION POWER FOR SFC

• Is there a fundamental limitation on operating pressure in SFC?



50cm, 2.1mm ID, 1.8µm at 1.3mL/min

 $2.7 \mu m$ poroshell, 3mm ID



R. De Pauw et al., J. Chromatogr. A 1374 (2014) 247–253

LIMITS OF SEPARATION POWER FOR SFC

- Bamba and co-workers in 2014 proposed wide elution gradients: SFC => EFLC => HPLC, i.e. from CO_2 to MeOH
- Elution of a wide range of analytes: least to most polar
- Referred to as "unified chromatography" (UC)



Nucleoshell HILIC 2.7 μ m, 100 × 3.0 mm column



K. Taguchi et al., J. Chromatogr. A 1362 (2014) 270–277 V. Desfontaine et al., J. Chromatogr. A 1562 (2018) 96–107

LIMITS OF SEPARATION POWER FOR SFC

• West and co-workers applied combined flow rate and backpressure gradients:



Fig. 1. Gradient profile in the optimized method showing the variation of co-solvent (blue lines), back-pressure (orange lines) and flow rate (green lines).

Chiralpak ZWIX(+), 150 × 3.0 mm, 3 μ m

A. Raimbault et. al., J. Chromatogr. A 1616 (2020) 460772



- Separation performance not only determined by mobile phase viscosity, columns and pump pressure
- Instrumental contributions to dispersion and pressure drop:
 Injection





Trends in Analytical Chemistry 119 (2019) 115619



Fig. 1. Representative chromatogram and peak variances for the estimation of system variance by using the "in presence of column" methodology. Instrument: Acquity UPC², column: Waters BEH 1.7 μ m (100 mm × 3.0 mm) mobile phase: 10% MeOH in CO₂, temperature: 40 °C, flow rate: 2 mL/min. Test analytes: butylparaben, caffeine, theobromine, betamethazone, chlorthalidone and hydrochlorothiazide.



A. Grand-Guillaume Perrenoud et al., J. Chromatogr. A 1314 (2013) 288–297

- $\sigma_{V,col}^2 = (V_0^2/N) \cdot (1+k_e)^2$
- How large is column dispersion: $(k_e = 2.16)$
 - 4.6mm ID, 25cm, 5μm: 2500 μL²
 - 4.6mm ID, 15cm, 3.5μm: 1000 μL²
 - 2.1mm ID, 10cm, 2 μm fully porous: 17 μL^2
 - 2.1mm ID, 5cm, 1.5μm superficially porous particles: 6 μL²
- Evolution in extra-column dispersion in LC
 - HPLC systems: 20-100 μL²
 - UHPLC systems: < 10 $\mu L^2,$ optimized systems < $2\mu L^2$
- Extra-column dispersion in SFC ?



Trends in Analytical Chemistry 119 (2019) 115619

$\sigma_{V,col}^2 = (V_0^2/N) \cdot (1+k_e)^2$

- Extra-column dispersion in SFC
 - Much less investigated than in LC
 - Estimated by Guillarme and co-workers (10% MeOH in CO₂)
 - 23 and 83 μL^2 for F = 1ml/min and 2ml/min on UPC²
 - 55 and 81 μ L² for F = 1ml/min and 2ml/min on Agilent SFC

$$\sigma_{V,tot}^{2} = \sigma_{V,ec}^{2} + \sigma_{V,col}^{2}$$

$$\int_{N_{obs}}^{N_{obs}} = (V_{0}^{2}/\sigma_{V,tot}^{2}) \cdot (1+k_{e})^{2}$$



A. Grand-Guillaume Perrenoud et al., J. Chromatogr. A 1314 (2013) 288–297

 $\sigma_{V,col}^2 = (V_0^2/N) \cdot (1+k_e)^2$ $\sigma_{V,tot}^2 = \sigma_{V,ec}^2 + \sigma_{V,col}^2$ $N_{obs} = (V_0^2/\sigma_{V,tot}^2) \cdot (1+k_e)^2$

- Extra-column dispersion in SFC
 - 50 μ L² extra-column dispersion for columns of 15cm long, with N = 25000, as a function of k_e for different column ID's
 - Observed efficiency



3.0 mm ID trade-off between ECBB and flow rate limitations





- Sources of extra-column dispersion in SFC
 - Analytes are dissolved in a strong(er) solvent (~mainly organic) than mobile phase (~ mainly CO₂)
 - Detector cells design and volumes not optimized as in UHPLC
 - Often larger ID tubing employed (170µm vs. 75-120µm in UHPLC)



- Sources of extra-column dispersion in SFC: detector
 - Detector contribution: SFC 1.7µL flow cell vs. ultra-low dispersion max light cartridge

Agilent 1290 Ultra-Low Dispersion Max-Light Cartridge Flow Cell; V(σ)=0.6µL



Max. Press.: 6 MPa/60 bar/870 psr



R. De Pauw et al., J. Chromatogr. A 1403 (2015) 132–137

- Sources of extra-column dispersion in SFC: detector
 - Sample solvent EtOH/IPA/hexane: 10/5/85, $V_{inj} = 0.8\mu L$
 - P_{back} = 130 bar, 2.1x100mm 1.8µm, F=1ml/min, 8 V% MeOH
 - Tubing
 - Length (50cm)
 - Inner diameter 250 µm ID





R. De Pauw et al., J. Chromatogr. A 1403 (2015) 132–137

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R. De Pauw et al., J. Chromatogr. A 1403 (2015) 132–137

- Sources of extra-column dispersion in SFC: tubing
 - Sample solvent EtOH/IPA/hexane: 10/5/85, $V_{inj} = 0.8\mu L$
 - P_{back} = 130 bar, 2.1x100mm 1.8µm, F=1ml/min, 8 V% MeOH
 - Tubing
 - Length (50cm)
 - Inner diameter variable
 - 250µm
 - 170µm





R. De Pauw et al., J. Chromatogr. A 1403 (2015) 132–137

- Sources of extra-column dispersion in SFC: tubing
 - Sample solvent EtOH/IPA/hexane: 10/5/85, $V_{inj} = 0.8\mu L$
 - P_{back} = 130 bar, 2.1x100mm 1.8µm, F=1ml/min, 8 V% MeOH
 - Tubing
 - Length (50cm)
 - Inner diameter variable
 - 250µm
 - 170µm
 - 120µm





R. De Pauw et al., J. Chromatogr. A 1403 (2015) 132–137

- Optimized system for minimal extra-column dispersion
 - Sample solvent EtOH/IPA/hexane: 10/5/85, $V_{inj} = 0.8\mu L$
 - P_{back} = 130 bar, 2.1x100mm 1.8µm, F=1ml/min, 8 V% MeOH
 - Tubing 50cm, 120µm





R. De Pauw et al., J. Chromatogr. A 1403 (2015) 132–137

EXTRA-COLUMN PRESSURE

- Watch out for narrow ID tubing in SFC at high flow rate!
- 4.6 mm ID column (10% MeOH) and 120 μ m tubing in system





R. De Pauw et al., J. Chromatogr. A 1361 (2014) 277–285

- Feed injector: dilute sample with mobile phase
- Injection flow rate can be varied (max. 1 ml/min)
- Overfeed volume needed to fully inject loaded sample volume





K. Vanderlinden et al., J. Chromatogr. A 1630 (2020) 461525

- Feed injector: dilute sample with mobile phase
- Injected sample plug increases in volume $\sim 1+F/F_{feed}$
- Sample solvent is diluted with mobile phase (~ mainly CO₂)





K. Vanderlinden et al., J. Chromatogr. A 1630 (2020) 461525

• Feed injector: effect of feed flow rate F_{feed} for F = 2.5 ml/min





K. Vanderlinden et al., J. Chromatogr. A 1630 (2020) 461525

- Optimal feed flow rate F_{feed} found for strong dilution (3-6x)
- Optimal ratio F_{feed}/F independent on flow rate F
- Stronger dilution for solutes with larger dependency of k on $\boldsymbol{\phi}$





K. Vanderlinden et al., J. Chromatogr. A 1630 (2020) 461525

- CO₂-based mobile phases exhibit interesting properties but are often overstated for modern conditions and applications
- Kinetic separation performance of modern SFC instruments is similar to UHPLC
- Need for backpressure and lower maximum pump pressure decreases potential gains from low viscosity mobile phase and application of UC-gradients
- Extra-column band broadening can still be significantly improved for SFC to fully exploit advantages for small superficially porous particles and narrow ID columns



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